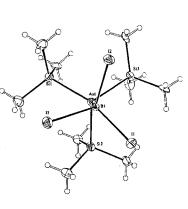
Main Group Chemistry, 1997, Vol. 2, pp. 85–88 Reprints available directly from the publisher Photocopying permitted by license only (b) 1997 OPA (Overseas Publishers Association) Amsterdam B.V. Published under license under the Gordon and Breach Science Publishers imprint. Printed in India.

## Index Abstracts

A low-temperature  $(-78^{\circ}\text{C})$  1:1 mole ratio reaction of BI<sub>3</sub> with As(SiMe<sub>3</sub>)<sub>3</sub> produced the 1:1 Lewis acid-base adduct I<sub>3</sub>B·As(SiMe<sub>3</sub>)<sub>3</sub> (1), whereas the identical reaction performed at room temperature (previously reported)<sup>1</sup> produced the dimeric compound, [I<sub>2</sub>BAs(SiMe<sub>3</sub>)<sub>2</sub>]<sub>2</sub> (2). Full characterization of compound **1** was conducted, including a single-crystal X-ray structural determination. The previously unreported X-ray crystal structures of **2** and Cl<sub>3</sub>B·As(SiMe<sub>3</sub>)<sub>3</sub> (**3**) have also been elucidated.

The unhindered 1,1-dimethyl-*N*-dimethylsilylsilanimine, generated under flash vacuum thermolysis conditions by "retroene" reaction, has been characterized for the first time by its photoelectron spectrum. The experimental ionization potentials (8.3 and 9.0 eV) respectively are associated with the ejection of an electron from the  $n_{\rm N}$  and  $\pi_{\rm Si=N}$  orbitals. An *ab initio* theoretical investigation provides further insight into the electronic and structural characteristics of the Si=N-Si backbone.





M. S. Lube, R. J. Jouet, R. L. Wells, P. S. White, and V. G. Young, Jr.

Further Investigations into the Synthesis and Characterization of Halo-Boron-Arsenic Compounds: X-Ray Crystal Structures of  $X_3B \cdot As(SiMe_3)_3$  (X = Cl, I) and  $[I_2BAs(SiMe_3)_2]_2$ 

89-96

V. Métail, S. Joanteguy, A. Chrostowska-Senio, G. Pfister-Guillouzo, A. Systermans and J. L. Ripoll

Gas-Phase Generation and Photoelectron Spectrum of 1,1-Dimethyl-*N*-Dimethylsilylsilanimine

97-106

Molecular Modelling of "spherical" cyclophosphazenic dandelion dendrimers shows that the whole moieties belong to the  $C_3$  point group, conversely to the starting material,  $N_3P_3Cl_6$ , which is of  $D_{3h}$  symmetry. Moreover, these spherical dendrimers have actually significant dipole moments due to a non-centrosymmetrical distribution of the lone pairs of the nitrogen atoms of their amino groups and functions.

Tetraphenylstibonium hydroxide (Ph<sub>4</sub>-SbOH) acted as a profitable base to effect the synthesis of  $\beta$ -lactams from haloamides. In most cases, acrylamides which were serious by-products in other studies formed in poor yields. Cyclic carbamates were also prepared under the same conditions from the corresponding halocarbamates.

The reaction of (Me<sub>3</sub>Si)MeNPCl<sub>2</sub>NSO<sub>2</sub>Cl with BCl<sub>3</sub> gives rise to the cyclic 2,2,4,4,6-pentachlor-3-methyl-6-oxo-1,6,2,3,5,4-oxathiaboradiazafosfinine-6-onio-2-ide Cl<sub>2</sub>PN(O)ClOBCl<sub>2</sub>NMe (**IV**) while the reaction of Me<sub>3</sub>SiNHPCl<sub>2</sub>NSO<sub>2</sub>Cl with BCl<sub>3</sub> leads to the linear compound bis(*N*-chlorosulfuryl-dichlorophosphazo-amino)chloroborane (ClSO<sub>2</sub>NPCl<sub>2</sub>NH)<sub>2</sub>-BCl (**VI**). Both compounds were characterized by <sup>31</sup>P- and <sup>11</sup>B-NMR spectrometry and elemental analysis, and their crystal structures were determined.

The hydrolytic condensation of PhSiCl<sub>3</sub> in acetone precipitates **5a** as the sole crystalline product in ~40% yield. The all-*cis* structure, which was originally assigned by J. F. Brown on the basis of IR spectroscopy and derivatization studies, was confirmed by a single-crystal X-ray diffraction study. Tetrasilanol **5a** can be exhaustively silylated with Me<sub>3</sub>SiCl, but all attempts to effect clean partial silylation produced complex mixtures of partially silylated T-resins.

J.-P. Fayet, F. Sournies, F. Crasnier, M.-C. Labarre and J.-F. Labarre

Why "Spherical" Cyclophosphazenic Dandelion Dendrimers Have a Dipole Moment?

107-110

M. Fujiwara, M. Tanaka, A. Baba, H. Ando, and Y. Souma

Novel Synthesis of Lactams and Cyclic Carbamates Using Tetraphenylstibonium Hydroxide

111-115

D. Novotný, J. Příhoda, Z. Žák, and J. Marek

Reactions of trichlorophosphazo-*N*-sulfurylchloride and its derivates *II*. The reaction of  $(Me_3Si)RNPCl_2NSO_2Cl$  (R = H, Me) with BCl<sub>3</sub>, the structures of cyclic Cl<sub>2</sub>PNS(O)ClOBCl<sub>2</sub>NMe and linear [ClSO<sub>2</sub>NPCl<sub>2</sub>NH]<sub>2</sub>BCl

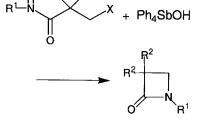
117-122

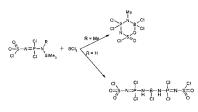
F. J. Feher, J. J. Schwab, D. Soulivong and J. W. Ziller

Synthesis, Characterization and Reactivity of *cis-cis-cis*- $[(C_6H_5)_4Si_4O_4(OH)_4]$ .

123--132

ОН





HO

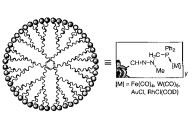
Ph

5a

The complexation properties of diphenylphosphino terminated dendrimers built from a cyclotriphosphazene core are studied. These compounds complex several types of organometallic moieties such as  $Fe(CO)_4$ ,  $W(CO)_5$ , RhCl(COD) and Au-Cl. Some of these reactions have been carried out up to the fifth generation (192  $Ph_2P \rightarrow Au$ -Cl terminal groups).

The reaction of Li{2,6-(NEt<sub>2</sub>CH<sub>2</sub>)<sub>2</sub>C<sub>6</sub>H<sub>3</sub>} with BCl<sub>3</sub>·Et<sub>2</sub>O in hexane yields BCl<sub>2</sub>[2-{BCl<sub>3</sub>(NEt<sub>2</sub>CH<sub>2</sub>)}-6-(NEt<sub>2</sub>CH<sub>2</sub>)C<sub>6</sub>H<sub>3</sub>] (1) as the thermodynamically favoured product. If BCl<sub>3</sub>·pyridine is employed, only Cl{2,6-(NEt<sub>2</sub>CH<sub>2</sub>)<sub>2</sub>C<sub>6</sub>H<sub>3</sub>} is formed. 1 reacts with [PPh<sub>4</sub>]Cl to give the BCl<sub>3</sub>-free product BCl<sub>2</sub>{2,6-(NEt<sub>2</sub>CH<sub>2</sub>)<sub>2</sub>C<sub>6</sub>H<sub>3</sub>} (2) with formation of [PPh<sub>4</sub>][BCl<sub>4</sub>]. Li{2-(NMe<sub>2</sub>CH<sub>2</sub>)C<sub>6</sub>H<sub>4</sub>} reacts with BCl<sub>3</sub> to give BCl<sub>2</sub>{2-(NMe<sub>2</sub>CH<sub>2</sub>)C<sub>6</sub>H<sub>4</sub>} (3). 1–3 were characterized spectroscopically (NMR, IR, MS) and their crystal structures were determined.

 $Se_3O_7$  reacts with excess pyridine to form the donor-acceptor (DA) complexes  $SeO_2$ -Py and  $Se_2O_5$ -5Py. The latter compound was characterized by elemental analysis, Raman spectroscopy, thermal analysis and X-ray crystallography.



M. Bardaji, M. Slany, M.-L. Lartigue, A.-M. Caminade, B. Chaudret and J. P. Majoral

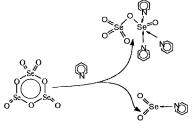
Complexation Properties of Bowl-shaped Dendrimers

133-140

R. Schlengermann, J. Sieler and E. Hey-Hawkins

Synthesis and Molecular Structures of the Base-stabilized Arylboron Dichlorides  $BCl_2$ - $\{2, 6-(NEt_2CH_2)_2C_6H_3\}$  and  $BCl_2$ - $\{2-(NMe_2CH_2)C_6H_4\}$ 



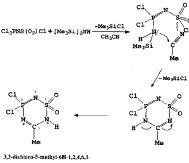


K. Neplechová, Z. Žák and J. Toužín

Reaction of Triselenium Heptoxide with Pyridine – Novel DA-complex  $4Py \cdot Se_2O_5$ 

149-153

Trichlorophosphazo-N-sulfurylchloride  $Cl_3PNS(O_2)Cl$  (1) reacts with  $[(CH_3)_3Si]_2NR$ , R = H,  $CH_3$  to give its N-trimethylsilyl derivatives  $(CH_3)_3SiN(H)P(Cl_2)NS(O_2)Cl$  (2) and  $(CH_3)_3SiN(CH_3)P(Cl_2)NS(O_2)Cl$  (3). At elevated temperatures (2) reacts with  $CH_3CN$  under an intramolecular elimination of  $(CH_3)_3SiCl$  and a formation of 3, 3-dichloro-5-methyl-6H-1, 2, 4, 6, 3thiatriazaphosphinine-1, 1-dioxide  $P(Cl_2)NS(O_2)N(H)C(CH_3)N$ .



3,3-dichloro-5-methyl-6H-1,2,4,6,3thlatriazaphosphinine-1,1-dioxide Z. Žák, A. Růžička, M. Vlčková and L. Fröhlichová

Reactions of Trichlorophosphazo-N-Sulfurylchloride and its Derivatives I. Synthesis of a New Heterocycle Containing P, N, S, and C Atoms in its Six-membered Ring. Crystal Structures of (CH<sub>3</sub>)<sub>3</sub>SiN(H)P-(Cl<sub>2</sub>)NS(O<sub>2</sub>)Cl, (CH<sub>3</sub>)<sub>3</sub>SiN-(CH<sub>3</sub>)P(Cl<sub>2</sub>)NS(O<sub>2</sub>)Cl, and P(Cl<sub>2</sub>)NS(O<sub>2</sub>)N(H)C(CH<sub>3</sub>)N at 150 K

155-160

N. Yanagihara, S. Tanikawa, N. Suzuki, M. Rivera and T. Ogura

Formation of Tetramethylene Sulfoxide via Oxidation of Tetrahydrothiophene by Copper(II)

161-163

Reaction of copper(II) *p*-toluenesulfonate with tetrahydrothydrothiophene (THT) forms a THT cation radical and a THT dication by stepwise one-electron oxidation by  $Cu^{2+}$ . Generation of a dimeric species of THT dication is proposed to explain the formation of tetramethylene sulfoxide (TMSO) upon hydrolysis of the dimeric THT dication with H<sub>2</sub>O.